



# Solar Photovoltaic System Engineering

The Theory, Technology and Practice of Solar Photovoltaic  
Systems for Electric Power Generation

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with Guest Lectures by  
Justin Hitchcock, Doug Hall and others

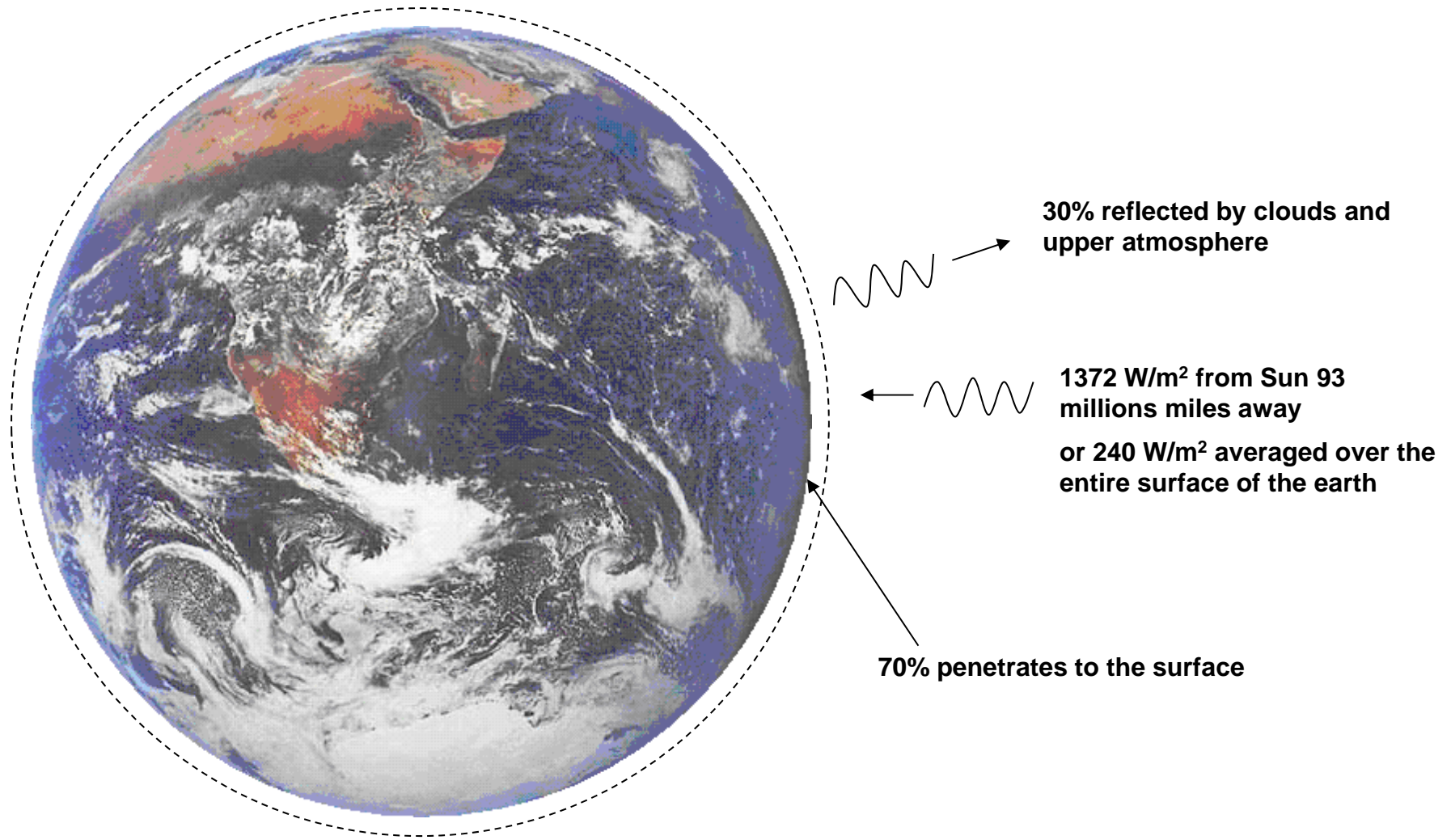
Lecture Material Credit:  
Arno Smits, UT Delft NV  
Dale Dolan, Cal Poly EE  
Gabe Davis and others at REC Solar  
Multiple Sources from Web, individually cited



**Week 1: The Solar Thermal Balance of the Earth  
And Causes of Global Warming**

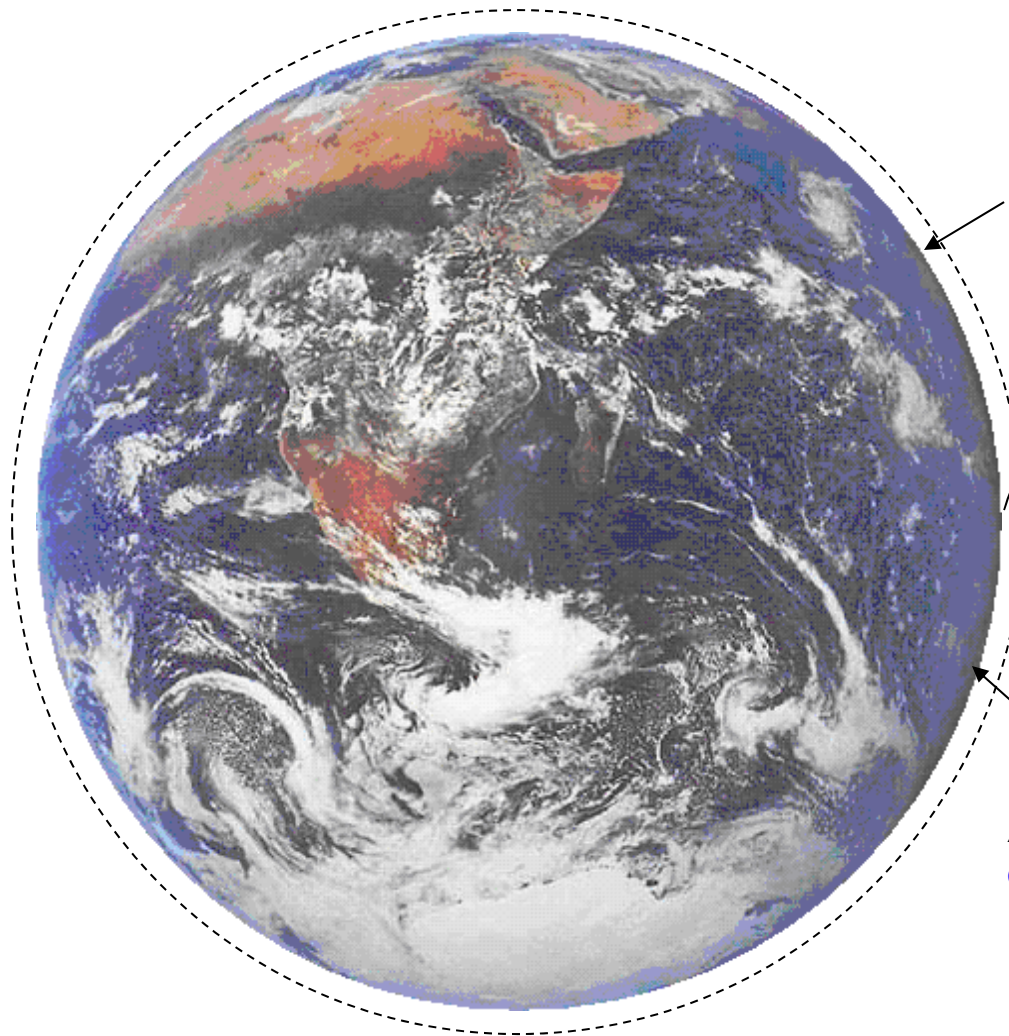
## Earth: A System in Delicate Balance

*The earth is a system in delicate climatic balance based on countless cycles and subsystems.*





## If no Greenhouse Gases....



Insolation reaching surface of earth, averaged over entire surface of earth: 240 watts/m<sup>2</sup>

All 240 watts/m<sup>2</sup> re-radiated because no greenhouse gases.

Average surface temperature = **-18 degrees C**  
or **0 degrees F**

*Life as we know it would not exist.*



## The Greenhouse Effect – Very Sensitive to Composition of Atmosphere

- Oxygen and nitrogen, the primary components of the atmosphere, are transparent to infrared energy, and do not block this re-radiation.
- But water vapor, methane, carbon dioxide, ozone, nitrous oxides, and synthesized gases such as chlorofluorocarbons absorb and re-radiate the reflected infrared energy both up into space, *and down back to the earth.*
- The retention of some of this heat near the surface is the *Greenhouse Effect.*
- This is a natural process critical to life on this planet. We rely on it to maintain an average surface temperature of about 58.3 degrees F.
- An increase in the Greenhouse effect leads to *Global Warming.*

## Relative impacts on atmospheric heat retention by GHG

Based on concentrations (ppb) adjusted for heat retention characteristics	% of Greenhouse Effect	% Natural	% Man-made
Water vapor	<b>95.000%</b>	94.999%	<b>0.001%</b>
Carbon Dioxide (CO <sub>2</sub> )	<b>3.618%</b>	3.502%	<b>0.117%</b>
Methane (CH <sub>4</sub> )	0.360%	0.294%	<b>0.066%</b>
Nitrous Oxide (N <sub>2</sub> O)	0.950%	0.903%	<b>0.047%</b>
Misc. gases ( CFC's, etc.)	0.072%	0.025%	<b>0.047%</b>
Total	100.00%	<b>99.72</b>	<b>0.28%</b>

Source: [http://www.geocraft.com/WVFossils/greenhouse\\_data.html](http://www.geocraft.com/WVFossils/greenhouse_data.html)

Sample contradictory arguments:  
<http://www.skepticalscience.com/empirical-evidence-for-co2-enhanced-greenhouse-effect.htm>



## Anthropogenic (man-made) vs naturally-occurring GHGs

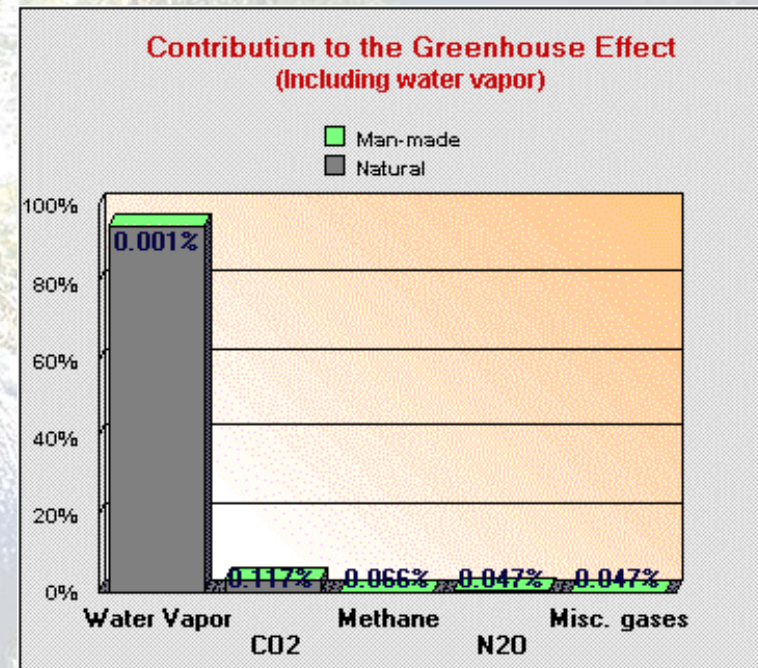
Naturally-occurring GHGs dominate the global temperature balance physics.

Water vapor is responsible for 95% of greenhouse effect, and is 99.9+% naturally-occurring.

Anthropogenic CO<sub>2</sub> contributions cause only about 0.117% of greenhouse effect, And some argue that this means that we can ignore this contribution.

Including all GHGs total human contribution to the greenhouse effect is around 0.28%. Therefore, some argue that this means that we can ignore human contributions.

But the thermal balance of the planet is *very* delicate, and the ramifications of a few degrees F temperature change on all forms of life is huge (especially for human society).





## The Dominant GHGs: Water vapor and Carbon Dioxide

- Both water vapor and Carbon dioxide are greenhouse gases. They both trap outgoing longwave infrared radiation in the troposphere.
- On a molar average basis, water vapor is about half as effective as a GHG compared with carbon dioxide. The exact effect varies highly with the form of the water vapor, e.g., humidity vs clouds.
- Water vapor in the atmosphere averages 2-3%, carbon dioxide averages 0.04%. (More than 60 times as much H<sub>2</sub>O as CO<sub>2</sub>.)
- Water vapor is removed from the atmosphere by precipitation. CO<sub>2</sub> is removed from the air by plant life, especially aquatic.
- The water vapor cycle is an unstable system: as global temperatures rise, more water is vaporized, leading to more heat entrapment and even higher temperatures. Water vapor feedback roughly doubles the amount of warming caused by CO<sub>2</sub>. And when all feedback loops are considered, the total warming effect due to a 1°C change caused by CO<sub>2</sub> is nearly 3°C.

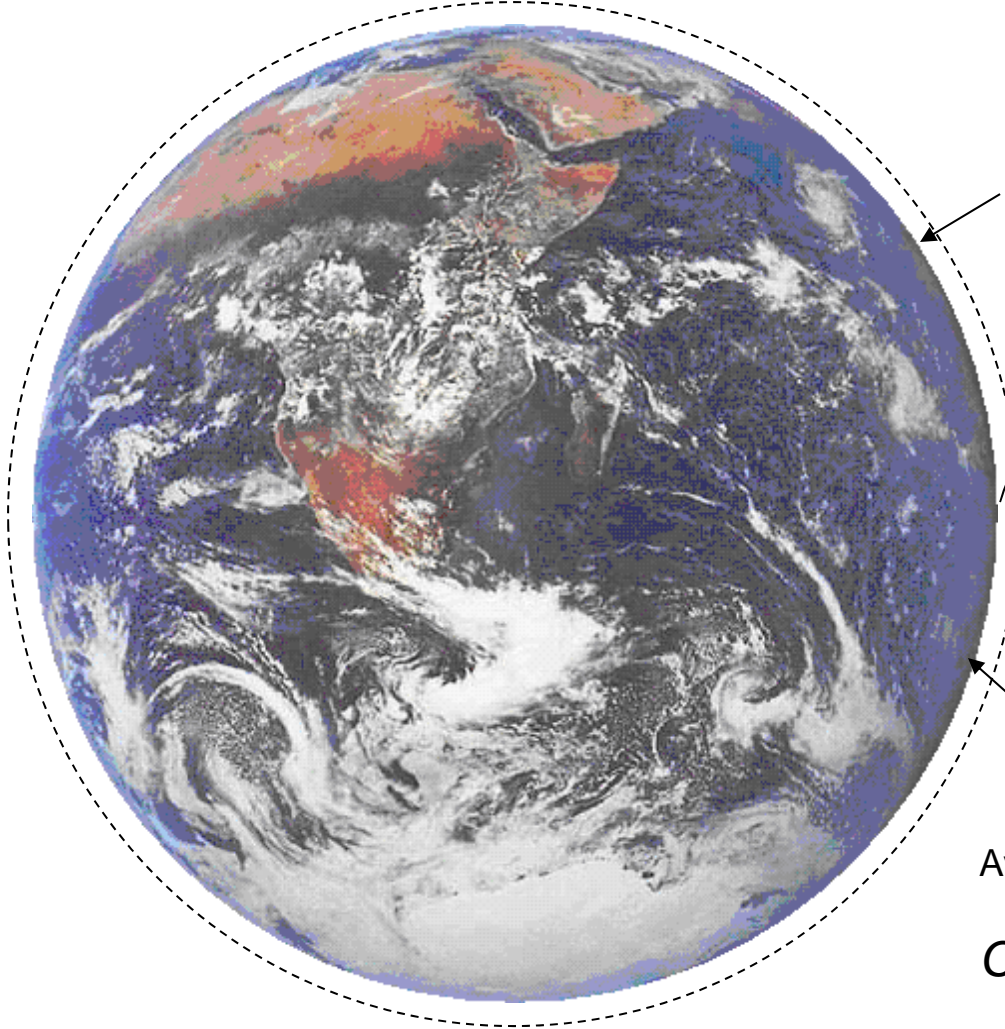


## What do we know vs what must we attempt to predict?

- The role of CO<sub>2</sub> as a greenhouse gas was first published in 1896 by Swedish scientist Svante Arrhenius.
- Approximately 100 years ago, before large-scale use of fossil fuels, the atmosphere contained an average of **275 ppm CO<sub>2</sub>**. Greenhouse gases intercepted 88% of the re-radiated energy, which yielded an equilibrium average surface temperature of **57 °F in 1900**.
- Today, as a direct result of the combustion of fossil fuels, the atmosphere contains slightly about **400 ppm CO<sub>2</sub>**. There has also been an equivalent reduction in oxygen content.
- The natural CO<sub>2</sub> re-absorption and conversion processes of the earth cannot keep up with our man-made contributions.
- But we *know* that the average surface temperature **rose  $1.33 \pm 0.32$  °F to 58.3 °F between 1900 and in 2005 (when CO<sub>2</sub> was 320 ppm)**

## With Naturally-occurring Greenhouse Gases....

**275 ppm CO<sub>2</sub>**



Insolation reaching surface of earth, averaged over entire surface of earth: 240 watts/m<sup>2</sup>

**12% (29 watts/m<sup>2</sup>) re-radiated** while 88% (211 watts/m<sup>2</sup>) absorbed by greenhouse gases.

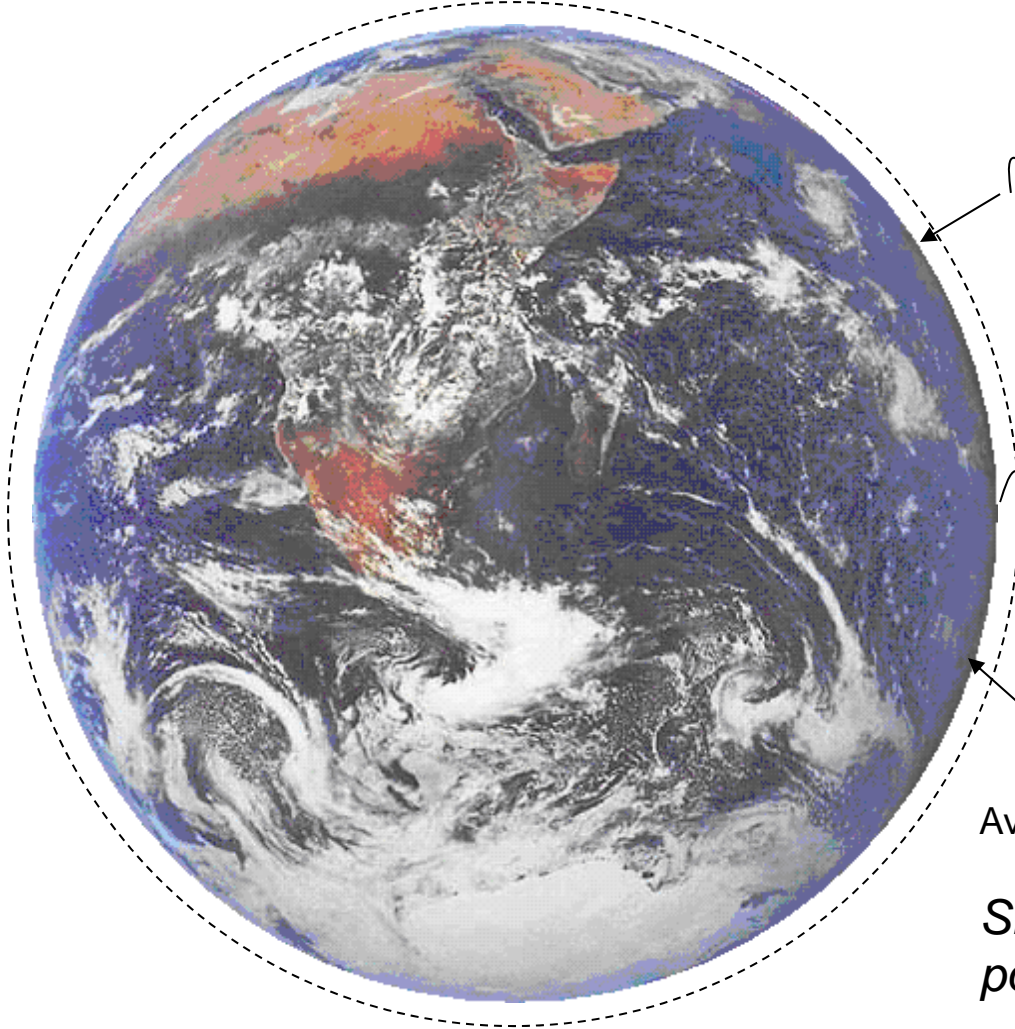
Average surface temperature = **57 degrees F**

*Comfortable for life, including humans*



After 100 years of burning fossil fuels....

**400 ppm CO<sub>2</sub>**



Insolation reaching surface of earth, averaged over entire surface of earth: 240 watts/m<sup>2</sup>

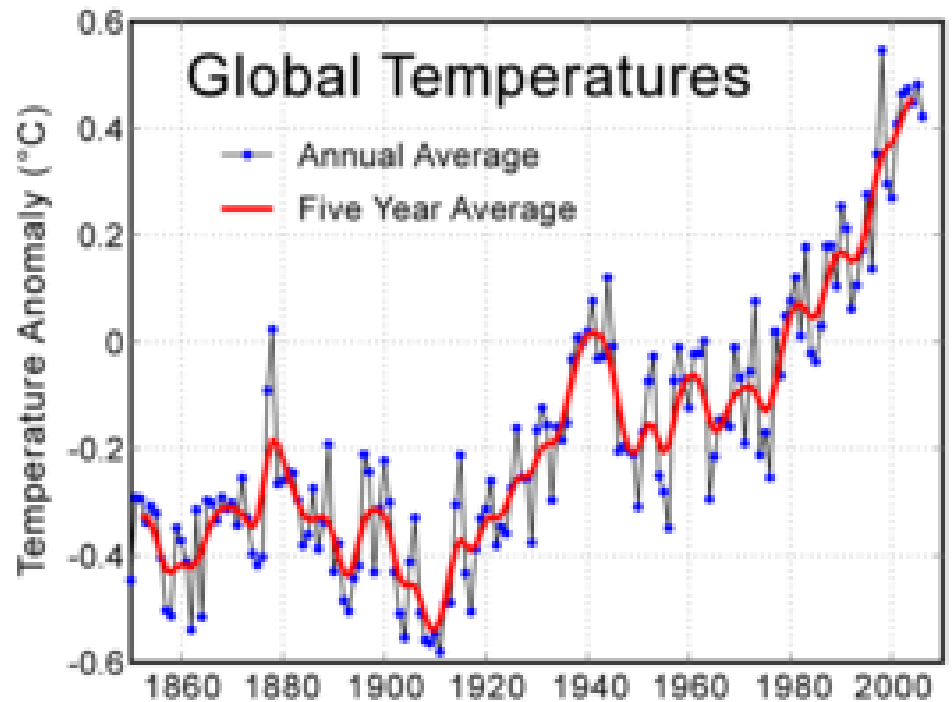
**10% (24 watts/m<sup>2</sup>) re-radiated** while >90% (216 watts/m<sup>2</sup>) absorbed by greenhouse gases.

Average surface temperature = **58.3 degrees F**

*Small change in temperature, but potentially large difference in climate.*

## Climate – Very Sensitive to the Heat Balance of the Planet

- If we continue to burn fossil fuels until they start to become exhausted, the CO<sub>2</sub> concentration is expected to increase to about 550 ppm.
- Climate model projections reported by the IPCC\* suggest that the average surface temperature will likely **rise** another 1.1 to 6.4 °C (**2.0 to 11.5 °F**) during the 21st century
- Ramifications for global climate change *could* be huge, as speculated in many forums, such as film by Al Gore “An Inconvenient Truth”. There is not a clear consensus on this.



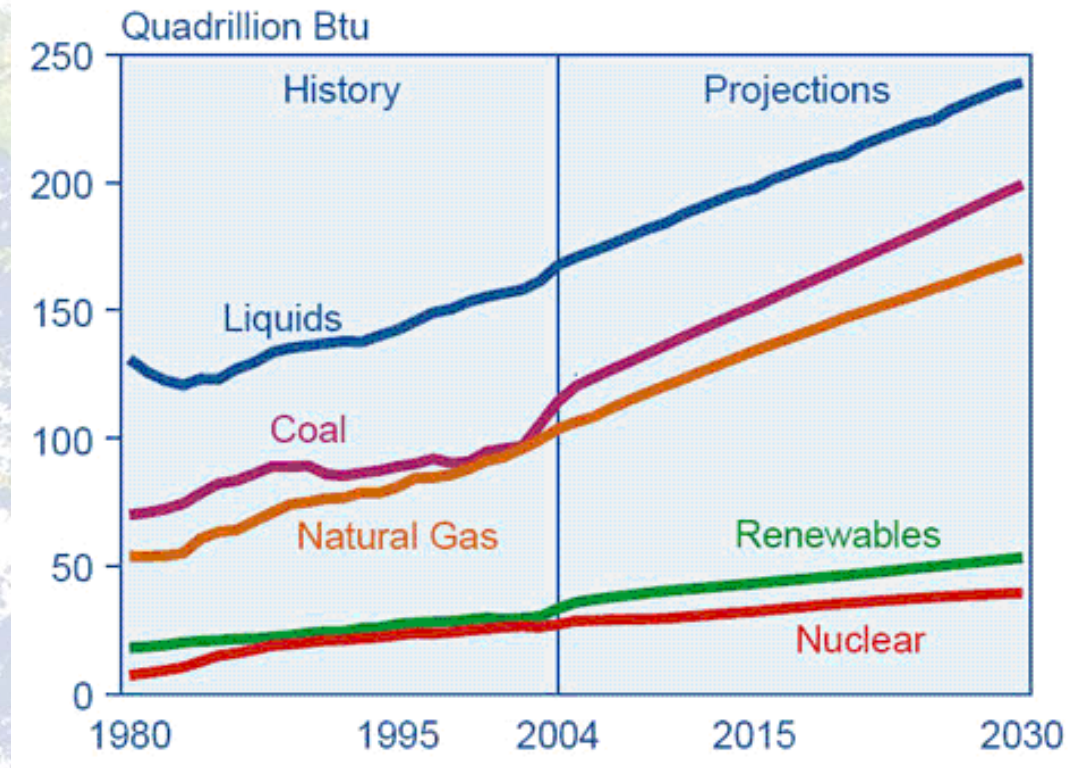
Sources: **History:** Energy Information Administration (EIA), *International Energy Annual 2004* (May-July 2006), web site [www.eia.doe.gov/iea](http://www.eia.doe.gov/iea). **Projections:** EIA, *System for the Analysis of Global Energy Markets* (2007).

\* Intergovernmental Panel on Climate Change  
<http://www.ipcc.ch/>



## Unexpected Contribution of Humans: Combustion of Fossil Fuels

- Fossil fuels (hydrocarbons or coal) contain organic carbon stored and reduced to its present form **over hundreds of millions of years**
- Combustion of fossil fuels using this long-stored carbon into the atmosphere
- We currently derive at least **86% of our global energy from combustion of fossil fuels**

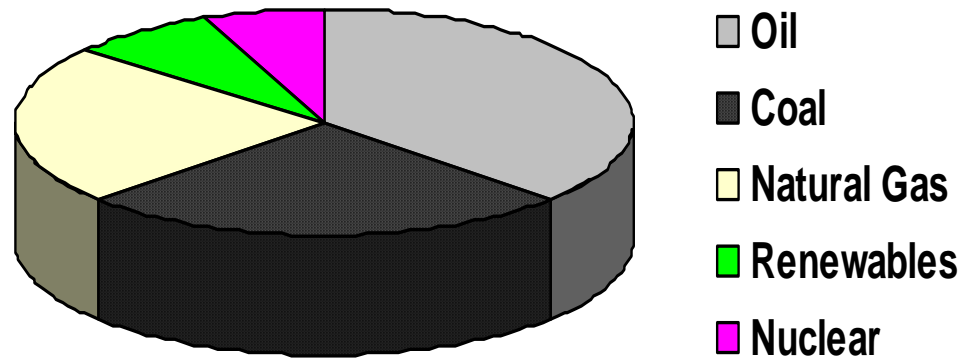


Sources: **History:** Energy Information Administration (EIA), *International Energy Annual 2004* (May-July 2006), web site [www.eia.doe.gov/iea](http://www.eia.doe.gov/iea). **Projections:** EIA, System for the Analysis of Global Energy Markets (2007).

# Fossil Fuel Combustion

## WORLD ENERGY CONSUMPTION

86% Fossil Fuels



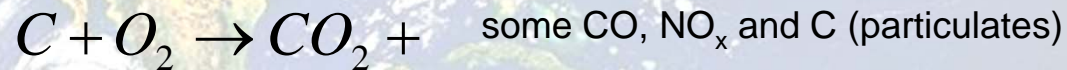
The International Energy Outlook 2007 (IEO2007)  
Energy Information Administration (EIA) – U.S. Government



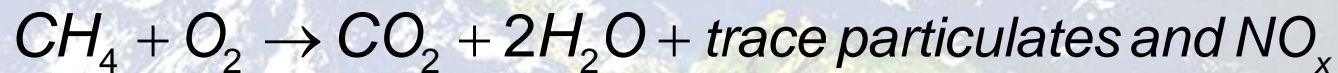
## Relative CO<sub>2</sub> Contributions and O<sub>2</sub> Utilization of Some Fuels:

Remember, all HC's burn in air to create predominantly CO<sub>2</sub> and H<sub>2</sub>O

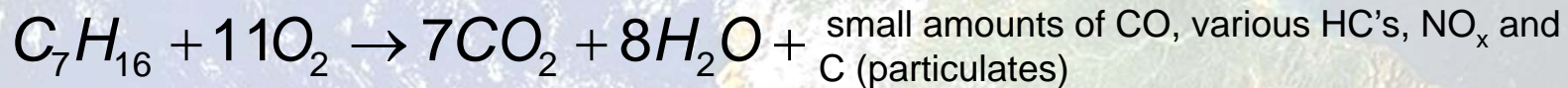
Combustion of Coal (no sulfur)



Combustion of Methane (majority component of natural gas)



Combustion of Gasoline (average composition represented by n-heptane C<sub>7</sub>H<sub>16</sub>)





## Sunlight to Energy: The Plant-based Carbon Cycle

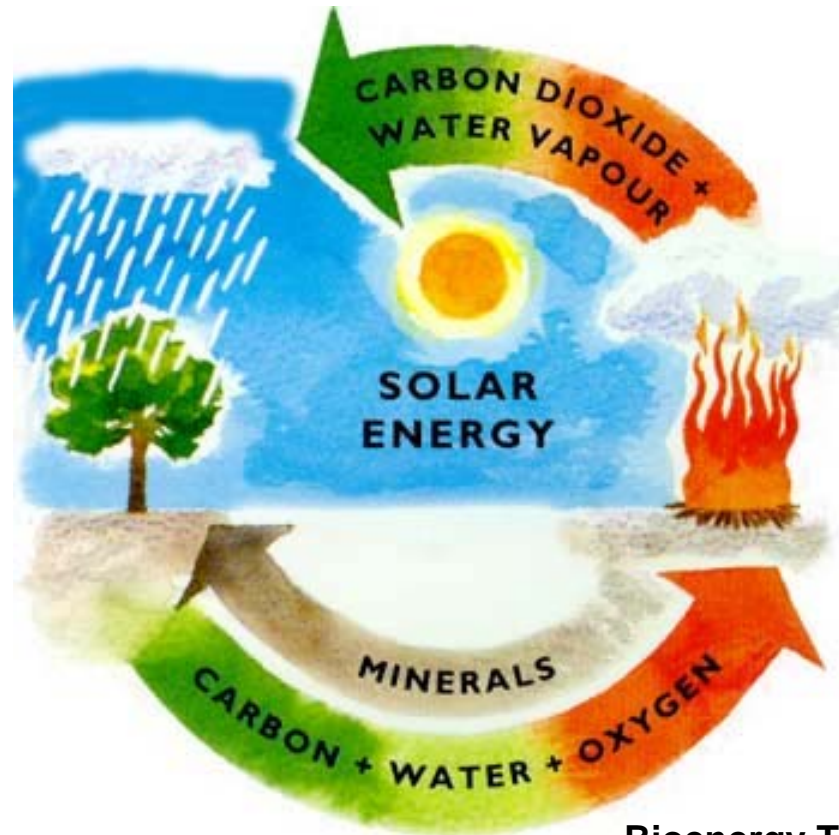
- Fossil fuels (hydrocarbons or coal) contain organic carbon stored from decayed plant life and reduced to its present form *over hundreds of millions of years*.
- Combustion of fossil fuels releases this stored carbon into the atmosphere, overloading the normal **Biological Carbon Cycle** of the planet, which has a time constant between *a year to a few decades*.
- The closed carbon cycle of agriculture and biomass-derived fuels **returns the carbon back to the fuel feedstock in a matter of years**, rather than the millions of years to took to make fossil fuels.





# Renewable Fuels

- Closed-loop, carbon neutral with a *short time constant*



Bioenergy Technology Ltd.



## Every potential solution has its costs

All “solutions” have consequences – some short term, and some long term.

**Technology is only one component...** every solution involves people, politics, and economics.

The profit base of much of an industrialized economy is predicated on transferring costs to the general public and the planet.

Examples:

- the complete environmental costs and energy balance for corn-derived ethanol
- the waste storage risk of nuclear power
- the environmental damage done by hydroelectric dams
- the size, noise and risks to avian life of wind generators
- the “true” life cycle costs of any manufactured product
- the energy and environmental consequences of replacing an inefficient machine with a more efficient one

*The real problem is much greater than obvious – threaded throughout global society, politics and economics.*





# Fossil-Fuel Combustion

**CO<sub>2</sub>**

**Air Pollution**

**Extraction**

**Habitat Loss**  
**More Pollution**

**Transport**

**Oil Spills**  
**More Pollution**

**Refining**

**More Pollution**



**Water Pollution**

**Hazardous Waste**

